



SSC JE Mains 2022



Mechanical Engineering

Most Expected
Questions

SET-10



UPCOMING

Double Validity

MECHANICAL ENGINEERING KA MAHAPACK

By Adda247

BILINGUAL

SUBSCRIPTION FEATURES

CIVIL MAHA PACK

All Live Classes	Yes
All Test Series	Yes
All Mock Video Solutions	Yes
All Revision Batches	Yes
All Recorded Videos	Yes
All eBooks	Yes
Personality Development	Yes
Spoken English	Yes

2021

Double Validity

BILINGUAL



JKSSB JE

Mechanical

Selection Batch

9 AM to 7 PM

BILINGUAL

Mission SSC JE 2023

Mechanical Engineering 2.0

Foundation of your Success



Start Jan 16, 2023

9 AM to 11 PM

Q Explain the boiler classification **15 marks**

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Q Explain the boiler classification 15 marks

Classification of boilers

1. Depending upon to the position of water and flue gases :
 - (a) Smoke tube or fire tube boilers →
 - (b) Water tube boilers →
2. Depending upon to the position furnace :
 - (a) Internally fired boilers
 - (b) Externally fired boilers
3. Depending upon the position of axis of the boiler:
 - (a) Vertical boilers
 - (b) Horizontal boilers

④ A/c to pressure L.P.B $P < 25 \text{ bar}$

M.P.B → $20 < P < 80$

H.P.B $P > 80 \text{ bar}$

Subcritical Boiler $180 \text{ to } 221.2 \text{ bar}$

Supercritical $P > 221.2 \text{ bar}$

14

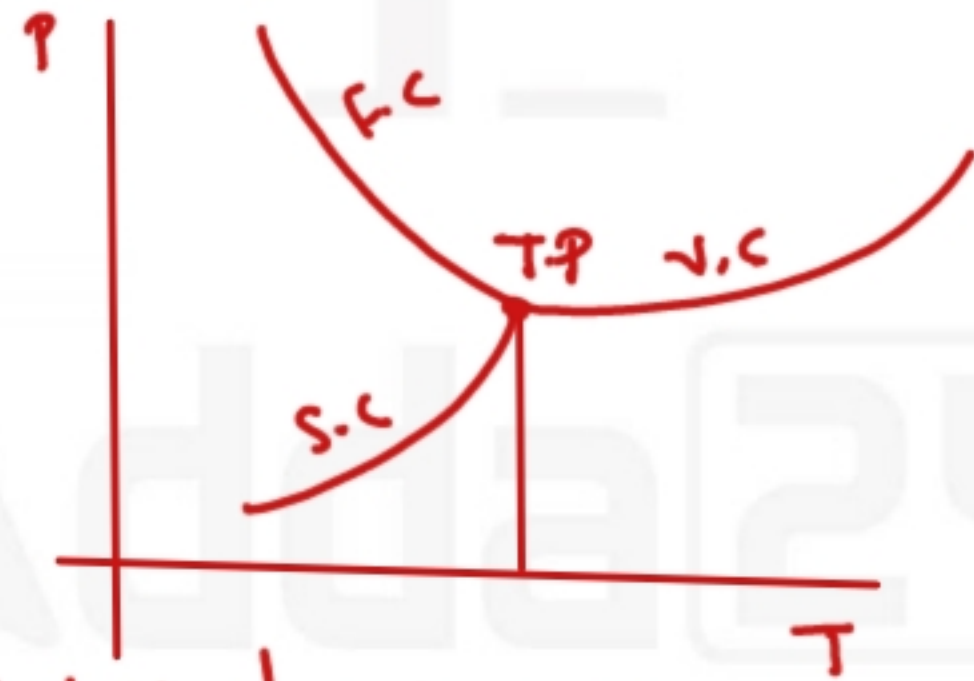
Q →

Difference

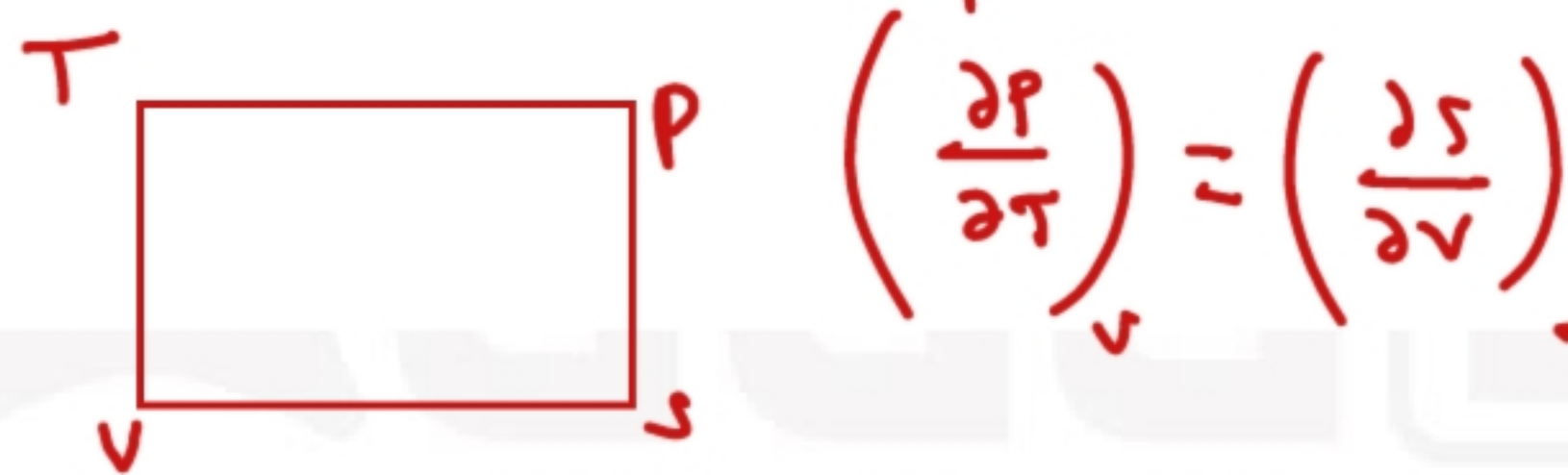
Water Tube Boiler	Fire tube Boiler
i) The water circulates inside the tubes which are surrounded by the hot flue gases from the furnace.	i) The hot gases from furnace passes through the tubes which are surrounded by water.
ii) It generates steam at a higher pressure up to 165 bar. 10 221 bar	ii) It can generate steam only up to 24.5 bar.
iii) The rate of generation of steam is high, up to 450 tons/hrs	iii) The rate of generation of steam is low, up to 9 tons/hrs
iv) Its overall efficiency 90%	iv) Its overall efficiency is only 75%
v) Its operating cost is high	v) Its operating cost is less
vi) The bursting chance of water tube boiler is more	vi) The bursting chance of fire tube boiler is less
vii) It is used for large power plant	vii) It is not suitable for large power plants
viii) <u>Examples:</u> <ul style="list-style-type: none">• Babcock and Wilcox boiler• Stirling boiler• La-Mont boiler• Benson boiler• Loeffler boiler• Yarrow boiler	viii) <u>Examples:</u> <ul style="list-style-type: none">• Simple vertical boiler• Cochran boiler• Lancashire boiler• Cornish boiler• Scotch boiler• Locomotive boiler• Velcon boiler

15 marks

Q The Clausius-Clapeyron Equation 15 marks



used to find LH during phase change at difference saturation pressure and temp



$$\frac{dp}{dT} = \frac{LH}{T_{sat}(v_2 - v_1)}$$

$$ds = \int \frac{dq}{T}$$

$$ds = \int_{T_{sat}} \frac{dq}{T_{sat}} = \frac{LH}{T_{sat}}$$

$$dp = \frac{LH dT}{T(v_2 - v_1)}$$

$$\int dp = \frac{LH}{(v_2 - v_1)} \int \frac{dT}{T}$$

$$(P_2 - P_1) = \frac{LH}{(v_2 - v_1)} \ln\left(\frac{T_2}{T_1}\right)$$

Q A system receives 50 kJ of heat while expanding with volume change of 0.14 m³ against an atmosphere of 1.2×10^5 N/m². A mass of 90 kg in the surroundings is also lifted through a distance of 5.5 m. (i) Calculate the change in energy of the system. (ii) The system is returned to its initial volume by an adiabatic process which requires 110 kJ of work. Find the change in energy of the system. (iii) For the combined processes of (i) and (ii), calculate the change in energy of the system **30 Marks**

Sol: $Q = 50 \text{ kJ}$ $\Delta V = 0.14$, $m = 90 \text{ kg}$ $P = 1.2 \times 10^5 \text{ N/m}^2$ $h = 5.5$

(i) $Q = (\Delta U) + W + E_p$

$S_0 = \Delta U + P(V_2 - V_1) + mgh$

$50 \times 10^3 = \Delta U + 1.2 \times 10^5 \times 0.14 + 90 \times 9.8 \times 5.5$

$(\Delta U) = 28.31 \times 10^3 \text{ Joule} = 28.31 \text{ kJ}$

(ii) $dQ = 0$ $Q = \Delta U + (\Delta P \cdot E) + W$

$\Delta U = -(\Delta P \cdot E + W) = -(mgh + 110 \times 10^3)$

$\Delta U = 86.82 \text{ kJ}$

$Q = \Delta U + \cancel{\Delta K} + \cancel{\Delta P \cdot E} + W$

