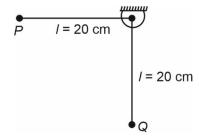
PHYSICS

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer:

Bob P is released from the position of rest at the moment shown. If it collides elastically with an identical bob Q hanging freely then velocity of Q, just after collision is $(g = 10 \text{ m/s}^2)$



- (1) 1 m/s
- (2) 4 m/s
- (3) 2 m/s
- (4) 8 m/s

Answer (3)

Sol. Velocity of *P* just before collision is = $\sqrt{2gl}$

= 2 m/sec

As collision is elastic and the mass of P and Q is equal therefore just after collision velocity of P is 0 and that of Q is 2 m/sec.

Choose the option showing the correct relation 2. between Poisson's ratio (σ), Bulk modulus (B) and modulus of rigidity (G).

(1)
$$\sigma = \frac{3B - 2G}{2G + 6B}$$
 (2) $\sigma = \frac{6B + 2G}{3B - 2G}$
(3) $\sigma = \frac{9BG}{3B + G}$ (4) $B = \frac{3\sigma - 3G}{6\sigma + 2G}$

(2)
$$\sigma = \frac{6B + 2G}{3B - 2G}$$

$$(3) \quad \sigma = \frac{9BG}{3B + G}$$

$$(4) \quad B = \frac{3\sigma - 3G}{6\sigma + 2G}$$

Answer (1)

Sol. $E = 2G(1 + \sigma)$

....(1)

 $E = 3B(1 - 2\sigma)$

....(2)

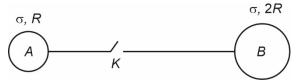
$$1 = \frac{2G}{3B} \left(\frac{1+\sigma}{1-2\sigma} \right)$$

 \Rightarrow 3B – 6B σ = 2G + 2G σ

 \Rightarrow 3B - 2G = σ (2G + 6B)

$$\sigma = \left(\frac{3B - 2G}{2G + 6B}\right)$$

Two conducting solid spheres (A & B) are placed at a very large distance with charge densities and radii as shown:



When the key K is closed, find the ratio of final charge densities.

- (1) 4:1
- (2) 1:2
- (3) 2:1
- (4) 1:4

Answer (3)

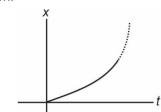
Sol. Final potential is same

$$\Rightarrow \frac{1}{4\pi\epsilon_0} \frac{Q_1}{R} = \frac{1}{4\pi\epsilon_0} \frac{Q_2}{2R} \quad(1)$$

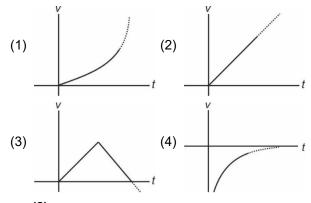
Also,
$$Q_1 + Q_2 = \sigma \cdot 4\pi R^2 + \sigma \cdot 4\pi (2R)^2$$
(2)

$$\Rightarrow \frac{\sigma_1}{\sigma_2} = 2.$$

Position-time graph for a particle is parabolic and is 4. as shown:



Choose the corresponding v - t graph



Answer (2)

Sol. Since $x \propto t^2$

$$\Rightarrow$$
 $V = \frac{dx}{dt} \propto t'$

⇒ Option 2 is correct

- 5. For a system undergoing isothermal process, heat energy is supplied to the system. Choose the option showing correct statements
 - (a) Internal energy will increase
 - (b) Internal energy will decrease
 - (c) Work done by system is positive
 - (d) Work done by system is negative
 - (e) Internal energy remains constant
 - (1) (a), (c), (e)
- (2) (b), (d)
- (3) (c), (e)
- (4) (a), (d), (e)

Sol. For isothermal process,

$$dT = 0$$

so, $dU = 0 \Rightarrow$ Internal energy remains same

$$dQ = dW$$

as dQ is positive,

so dW is positive

- 6. The heat passing through the cross-section of a conductor, varies with time 't' as $Q(t) = \alpha t \beta t^2 + \gamma t^3$. (α , β and γ are positive constants.) The minimum heat current through the conductor is
 - (1) $\alpha \frac{\beta^2}{2\gamma}$
- (2) $\alpha \frac{\beta^2}{3\gamma}$
- (3) $\alpha \frac{\beta^2}{\gamma}$
- (4) $\alpha \frac{3\beta^2}{\gamma}$

Answer (2)

Sol. Heat through cross section of rod

$$Q = \alpha t - \beta t^2 + \gamma t^3$$

so heat current = $\frac{dQ}{dt}$

heat current $=\frac{dQ}{dt}=\alpha-2\beta t+3\gamma t^2$

for heat current to be minimum

$$\frac{d^2Q}{dt^2} = -2\beta + 6\gamma t = 0$$

$$t = \frac{2\beta}{6\gamma} = \left(\frac{\beta}{3\gamma}\right)$$

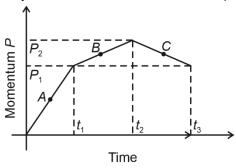
so minimum heat current

$$\left. \frac{\textit{dQ}}{\textit{dt}} \right|_{minimum} = \alpha - 2\beta \times \frac{\beta}{3\gamma} + 3\gamma \times \frac{\beta^2}{9\gamma^2}$$

$$=\alpha-\frac{2\beta^2}{3\gamma}+\frac{\beta^2}{3\gamma}$$

$$=\left(\alpha-\frac{\beta^2}{3\gamma}\right)$$

7. Momentum-time graph of an object moving along a straight line is as shown in figure. If $(P_2 - P_1) < P_1$ and $(t_2 - t_1) = t_1 < (t_3 - t_2)$ then at which points among A, B and C the magnitude of force experienced by the object is maximum and minimum respectively.



- (1) A, B
- (2) A, C
- (3) B, C
- (4) B, A

Answer (2)

Sol. *P P P A*

$$F_A = \frac{P_1}{t_1}$$

$$F_{B} = \frac{P_{2} - P_{1}}{t_{2} - t_{1}}$$

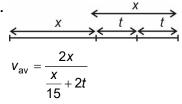
$$F_{\rm C} = \frac{P_2 - P_1}{t_3 - t_2}$$

Therefore the maximum force is at *A* and minimum force is at *C*.

- 8. A particle moving in unidirectional motion travels half of the total distance with a constant speed of 15 m/s. Now first half of the journey time it travels at 10 m/s and second half of the remaining journey time it travels at 5 m/s. Average speed of the particle is
 - (1) 12 m/s
- (2) 10 m/s
- (3) 7 m/s
- (4) 9 m/s

Answer (2)

Sol.

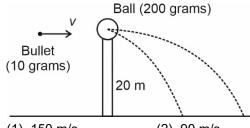


$$=\frac{2x}{\frac{x}{15}+\frac{2x}{10+5}}$$

= 10 m/sec

 A bullet strikes a stationary ball kept at a height as shown. After collision, range of bullet is 120 m and that of ball is 30 m. Find initial speed of bullet. Collision is along horizontal direction.

Take $g = 10 \text{ m/s}^2$



- (1) 150 m/s
- (2) 90 m/s
- (3) 240 m/s
- (4) 360 m/s

Answer (4)

Sol.
$$m_1V + m_2(O) = m_1v_1' + m_2V_2'$$
 ...(1)

$$\Delta t = \sqrt{\frac{2h}{g}} = 2s \quad ...(2)$$

$$\Rightarrow v_1' = \frac{120 \text{ m}}{2s} = 60 \text{ m/s}$$

&
$$v_2' = \frac{30 \text{ m}}{2 \text{ s}} = 15 \text{ m/s}$$

 \Rightarrow v = 360 m/s

10. If an inductor with inductive reactance, $X_L = R$ is connected in series with resistor R across an A.C voltage, power factor comes out to be P_1 . Now, if a capacitor with capacitive reactance, $X_C = R$ is also connected in series with inductor and resistor in the

same circuit, power factor becomes P_2 . Find $\frac{P_1}{P_2}$

- (1) $\sqrt{2}:1$
- (2) $1:\sqrt{2}$
- (3) 1:1
- (4) 1:2

Answer (2)

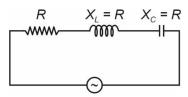
Sol.

$$Z = \sqrt{R^2 + R^2}$$

$$= \sqrt{2}R$$

$$P_1 = \cos\phi = \text{power factor} = \frac{R}{Z} = \left(\frac{1}{\sqrt{2}}\right)$$

When capacitor is also connected in series



The LCR circuit is in resonance stage

So,
$$Z = \sqrt{R^2 + (X_L - X_C)^2}$$

Z = F

$$P_2 = \cos\phi = \text{power factor } = \frac{R}{Z} = \frac{R}{R} = 1$$

So,
$$\frac{P_1}{P_2} = \frac{\left(\frac{1}{\sqrt{2}}\right)}{1} = \frac{1}{\sqrt{2}}$$

- Electromagnetic wave beam of power 20 mW is incident on a perfectly absorbing body for 300 ns. The total momentum transferred by the beam to the body is equal to
 - (1) $2 \times 10^{-17} \text{ Ns}$
- (2) $1 \times 10^{-17} \text{ Ns}$
- (3) $3 \times 10^{-17} \text{ Ns}$
- (4) $5 \times 10^{-17} \text{ Ns}$

Answer (1)

Sol. Total energy incident = *Pt*

So total initial momentum = $\frac{Pt}{c}$

Total final momentum = 0

Total momentum transferred = $\frac{Pt}{c}$

$$=\frac{20\times10^{-3}\times300\times10^{-9}}{3\times10^{8}}$$

$$= 2 \times 10^{-17} \text{ Ns}$$

- 12. The velocity of an electron in the seventh orbit of hydrogen-like atom is 3.6 × 10⁶ m/s. Find the velocity of the electron in the 3rd orbit.
 - $(1) 4.2 \times 10^6 \text{ m/s}$
- (2) 8.4×10^6 m/s
- $(3) 2.1 \times 10^6 \text{ m/s}$
- $(4) 3.6 \times 10^6 \text{ m/s}$

Answer (2)

Sol. For hydrogen like atom,

$$V \propto \frac{1}{n}$$

$$\left(\frac{v_1}{v_2}\right) = \left(\frac{n_2}{n_1}\right)$$

$$\Rightarrow \frac{3.6 \times 10^6}{v_2} = \frac{3}{7}$$

$$\Rightarrow v_2 = \frac{7}{3} \times 3.6 \times 10^6$$

$$= 8.4 \times 10^6 \text{ m/s}$$

13. Electric field in a region is given by $\vec{E} = \frac{a}{x^2}\hat{i} + \frac{b}{v^3}\hat{j}$,

where x & y are co-ordinates. Find SI units of a & b.

- (1) $a Nm^2C^{-1}$
- (2) $a Nm^3C^{-1}$
- $b Nm^3C^{-1}$
- $b Nm^2C^{-1}$
- (3) $a NmC^{-1}$
- (4) $a Nm^2C^{-1}$
- $b Nm^2C^{-1}$
- $b Nm^2C^{-1}$

Answer (1)

Sol. *E* – NC⁻¹

$$x^2 - m^2$$

$$v^3 - m^3$$

$$\Rightarrow$$
 a – Nm²C⁻¹

&
$$b - Nm^3C^{-1}$$

- 14. Coil A of radius 10 cm has N_A number of turns and I_A current is flowing through it. Coil B of radius 20 cm has N_B number of turns and I_B current is flowing through it. If magnetic dipole moment of both the coils is same then
 - $(1) I_A N_A = 4I_B N_B$
- (2) $I_A N_A = \frac{1}{4} I_B N_B$
- (3) $I_A N_A = 2I_B N_B$ (4) $I_A N_A = \frac{1}{2} I_B N_B$

Answer (1)

Sol. Magnetic dipole moment $\mu = NIA = NI\pi R^2$

So
$$\frac{\mu_A}{\mu_B} = \frac{N_A I_A R_A^2}{N_B I_B R_B^2} = 1$$

$$\frac{N_A I_A (10^2)}{N_B I_B (20^2)} = 1$$

$$N_A I_A = 4 N_B I_B$$

- 15. An ideal gas undergoes a thermodynamic process following the relation PT^2 = constant. Assuming symbols have their usual meaning then volume expansion coefficient of the gas is equal to
 - (1) $\frac{2}{\tau}$

Answer (2)

Sol. Volume expansion coefficient = $\frac{dV}{VdT}$

For PT^2 = constant

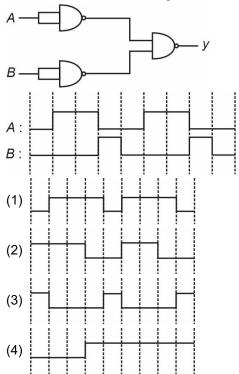
Or
$$\frac{T^3}{V}$$
 = constant

Or
$$\frac{dV}{dT} = (C) 3T^2$$

Or
$$\frac{dV}{VdT} = \frac{3T^2}{T^3}$$

$$\frac{dV}{VdT} = \frac{3}{T}$$

16. Consider a combination of gates as shown:



Answer (1)

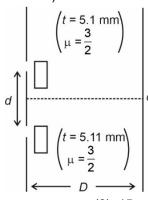
Sol. y = (A'B') = A + B

- ⇒ OR gate
- ⇒ Option 1
- 17. For the given YDSE setup. Find the number of fringes by which the central maxima gets shifted from point O.

(Given d = 1 mm

$$D = 1 \text{ m}$$

$$\lambda = 5000 \text{ Å}$$



(1) 10

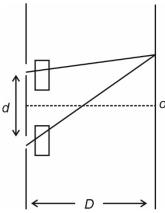
(2) 15

(3) 8

(4) 12

Answer (1)

Sol.



at central position, path difference, is,

$$(\mu - 1)t_1 - (\mu - 1)t_2$$

$$\Delta x = (\mu - 1) (t_1 - t_2)$$

$$\Delta x = \left(\frac{3}{2} - 1\right) (5.11 - 5.10) \,\mathrm{mm}$$

$$=\frac{1}{2}\times(0.01)$$
 mm

= 0.005 mm

$$= 5 \times 10^{-6} \text{ m}$$

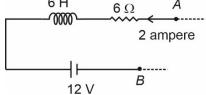
No. of fringes shifted =
$$\frac{\Delta x}{\lambda} = \frac{5 \times 10^{-6} \text{ m}}{5 \times 10^{-7} \text{ m}}$$

= 10

SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a **NUMERICAL VALUE.** For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 06.25, 07.00, -00.33, -00.30, 30.27, -27.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

21. In a part of a circuit shown:



Find $V_A - V_B$ in volts. It is given that current is decreasing at a rate of 1 ampere/s.

Answer (18)

Sol.
$$V_A - iR - L\frac{di}{dt} - 12 = V_B$$

$$\Rightarrow$$
 $V_A - V_B = +18$ volts

22. A particle undergoing SHM follows the position-time equation given as $x = A \sin\left(\omega t + \frac{\pi}{3}\right)$. If the SHM motion has a time period of T, then velocity will be maximum at time $t = \frac{T}{\beta}$ for first time after t = 0. Value of β is equal to

Answer (03.00)

Sol.
$$x = A \sin\left(\omega t + \frac{\pi}{3}\right)$$

$$\Rightarrow v = A\omega\cos\left(\omega t + \frac{\pi}{3}\right)$$

For maximum value of v

$$\cos\left(\omega t + \frac{\pi}{3}\right) = \pm 1$$

$$\Rightarrow \omega t + \frac{\pi}{3} = \pi$$
 (for nearest value of t)

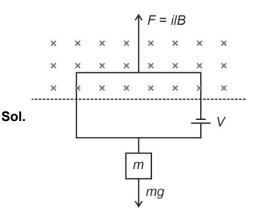
$$\omega t = \frac{2\pi}{3}$$

$$t=\frac{T}{3}$$

So
$$\beta = 3$$

23. A block of mass 1 g is equilibrium with the help of a current carrying square loop which is partially lying in constant magnetic field (B) as shown. Resistance of the loop is 10 Ω . Find the voltage (V) (in volts) of the battery in the loop.

Answer (10.00)



$$ilB = mg$$

$$i = \left(\frac{mg}{IB}\right) = \frac{(1 \times 10^{-3} \text{ kg}) \times (10 \text{ m/s}^2)}{(0.1 \text{ m}) \times (0.1 \text{ T})}$$

$$= 1 \times 10^{-3} \times 10^{3}$$

$$i = 1 A$$

As resistance of loop = 10 Ω

$$i = \frac{V}{R} = 1 \text{ A}$$

$$V = (1 \times 10) V$$

24. Initial volume of 1 mole of a monoatomic gas is 2 litres. It is expanded isothermally to a volume of 6 litres. Change in internal energy is *xR*. Find *x*.

Answer (00)

Sol.
$$\Delta U = nC_V \Delta T$$

= $nC_V(0)$ (: isothermal)
 $\Rightarrow \Delta U = 0$

25. An object is placed at a distance of 40 cm from the pole of a converging mirror. The image is formed at a distance of 120 cm from the mirror on the same side. If the focal length is measured with a scale where each 1 cm has 20 equal divisions. If the fractional error in the measurement of focal length is $\frac{1}{10 \ k}$ Find k.

Answer (60.00)

Sol.
$$u = -40 \text{ cm}$$

 $v = -120 \text{ cm}$
 $\frac{1}{V} + \frac{1}{U} = \frac{1}{f}$

$$\Rightarrow -\frac{1}{120} - \frac{1}{40} = \frac{1}{f}$$

$$\frac{1}{f} = \left(\frac{-1-3}{120}\right) = -\frac{4}{120}$$

$$f = -30 \text{ cm}$$

Least count of scale = $\left(\frac{1}{20}\right)$ cm

Fractional error =
$$\left(\frac{1}{20}\right) = \left(\frac{1}{600}\right)$$

as
$$\frac{1}{10 \ k} = \frac{1}{600}$$

$$k = 60$$

26. $\begin{array}{c|c}
1 \Omega & J_1 \\
\hline
 & & \\
2 \Omega & 5 V \\
\hline
\end{array}$

In two circuit shown above the value of current I_1 (in amperes) is equal to $-\frac{y}{5}$ A . Value of y is equal to

Answer (11.00)

Sol. $\begin{array}{c|c}
1 \Omega & I_3 & I_1 \\
\hline
 & I_1 + I_3 & \\
2 \Omega & I_2 & 5 V & 1 \Omega
\end{array}$

Using Kirchoff's law.

$$I_1 + I_3 - I_2 = -2$$
 ...(i)

$$I_3 + 2I_2 = 5$$
 ...(ii)

$$2I_2 - (I_3 - I_2) - (I_1 + I_3 - I_2) = 5$$
 ...(iii)

$$\Rightarrow I_1 = -\frac{11}{5} A$$

$$\Rightarrow y = 11$$

CHEMISTRY

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer:

- Caprolactam when heated at high temperature, gives
 - (1) Nylon 6, 6
 - (2) Dacron
 - (3) Teflon
 - (4) Nylon 6

Answer (4)

- **Sol.** Caprolactam on heating at high temperature gives Nylon-6 polymer.
- Molarity of CO₂ in soft drink is 0.01 M. The volume of soft drink is 300 mL. Mass of CO₂ in soft drink is
 - (1) 0.132 g
 - (2) 0.481 g
 - (3) 0.312 g
 - (4) 0.190 g

Answer (1)

Sol. Moles = $0.01 \times 0.3 = 0.003$

Mass = $0.003 \times 44 = 0.132$ gm

- 3. During the qualitative analysis of SO_3^{-2} using dilute H_2SO_4 , SO_2 gas evolved which turns $K_2Cr_2O_7$ solution (acidified H_2SO_4)
 - (1) Green
- (2) Black
- (3) Blue
- (4) Red

Answer (1)

Sol. Orange colour of dichromate solution (K₂Cr₂O₇) converts to green (Cr³⁺).

4. Number of lone pair of electrons on central atom?

	Column-I		Column-II
(A)	IF ₇	(P)	0
(B)	ICI ₄ -	(Q)	1
(C)	XeF ₂	(R)	2
(D)	XeF ₆	(S)	3

Match the following

- (1) $(A)\rightarrow(P)$; $(B)\rightarrow(Q)$; $(C)\rightarrow(R)$; $(D)\rightarrow(S)$
- (2) $(A)\rightarrow(P)$; $(B)\rightarrow(R)$; $(C)\rightarrow(S)$; $(D)\rightarrow(Q)$
- (3) $(A) \rightarrow (R)$; $(B) \rightarrow (S)$; $(C) \rightarrow (P)$; $(D) \rightarrow (Q)$
- (4) $(A)\rightarrow(S)$; $(B)\rightarrow(R)$; $(C)\rightarrow(Q)$; $(D)\rightarrow(P)$

Answer (2)

Sol. Molecule/species No. of lone pair

 $\begin{array}{ccc} \text{IF}_7 & \rightarrow 0 \\ \text{ICI}_4 & \rightarrow 2 \\ \text{XeF}_2 & \rightarrow 3 \\ \text{XeF}_6 & \rightarrow 1 \end{array}$

- 5. Which one of the following is water soluble?
 - (a) BeSO₄
 - (b) MgSO₄
 - (c) CaSO₄
 - (d) SrSO₄
 - (e) BaSO₄
 - (1) Only a and b
- (2) Only a, b, c
- (3) Only d and e
- (4) Only a and e

Answer (1)

- **Sol.** Solubility of sulphates of group-2 elements decreases down the group. BeSO₄ and MgSO₄ are appreciably soluble in water. CaSO₄, SrSO₄ and BaSO₄ are practically insoluble in water.
- 6. Shape of OF2 molecule is?
 - (1) Bent
- (2) Linear
- (3) Tetrahedral
- (4) T-shaped

Answer (1)

Sol.



It is sp^3 hybridised therefore its shape will be bent or V-shaped.

- 7. Inhibitor of cancer growth
 - (1) Cisplatin
 - (2) EDTA
 - (3) Cobalt
 - (4) Ethane 1, 2 diamine

Answer (1)

Sol. Cisplatin acts as an anticancer agent.

- 8. Speed of e⁻ in 7th orbit is 3.6×10^6 m/s then find the speed in 3rd orbit
 - (1) 3.6×10^6 m/s
 - (2) 8.4×10^6 m/s
 - (3) 7.5×10^6 m/s
 - (4) 1.8×10^6 m/s

Answer (2)

Sol. Speed of electron in nth orbit of a Bohr atom is given

$$v_n = (v_1)_H \frac{Z}{n}$$

If
$$n = 7$$

$$v_7 = (v_1)_H \frac{Z}{7} = 3.6 \times 10^6 \text{ m/s}$$

If
$$n = 3$$

$$v_3 = (v_1)_H \frac{Z}{3}$$

$$=\frac{7\times3.6\times10^6}{3}$$

$$= 8.4 \times 10^6 \text{ m/s}$$

9. Match the following:

Atomic Number

(i) 52

(p) s-block

(ii) 37

(q) p-block

(iii) 65

(r) d-block

(iv) 74

- (s) f-block
- (1) (i) \rightarrow (q); (ii) \rightarrow (p); (iii) \rightarrow (r); (iv) \rightarrow (s)
- (2) (i) \rightarrow (q); (ii) \rightarrow (p); (iii) \rightarrow (s); (iv) \rightarrow (r)
- (3) (i) \rightarrow (s); (ii) \rightarrow (r); (iii) \rightarrow (p); (iv) \rightarrow (q)
- (4) (i) \rightarrow (r); (ii) \rightarrow (p); (iii) \rightarrow (q); (iv) \rightarrow (s)

Answer (2)

- **Sol.** 37 is Rubidium belonging to 1st group of s-block.
- 10. Consider the following reactions

$$NO_2 \xrightarrow{UV} A + B$$

$$A + O_2 \longrightarrow C$$

$$B + C \longrightarrow NO_2 + O_2$$

- A, B and C are respectively
- (1) O, NO, O₃
- (2) NO, O, O₃
- (3) NO, O₃, O
- (4) O₃, O, NO

Answer (1)

Sol.
$$NO_2 \xrightarrow{UV} NO + O_{(B)} (A)$$

$$O + O_2 \longrightarrow O_3(C)$$

$$NO + O_3 \longrightarrow NO_2 + O_2$$

- 11. Which of the following option contains the correct match:
 - (List-I) (Reactions)
- (List-II) (Products)
- (A) Wurtz
- $(P) \langle O \rangle \langle O \rangle$
- (B) Fittig
- (Q) R R
- (C) Wurtz Fittig
- $(R) \langle O \rangle R$
- (D) Sandmeyer
- (S) (O) C
- (1) $A \rightarrow Q$; $B \rightarrow P$; $C \rightarrow R$; $D \rightarrow S$
- (2) $A \rightarrow P$; $B \rightarrow Q$; $C \rightarrow R$; $D \rightarrow S$
- (3) $A \rightarrow S$; $B \rightarrow R$; $C \rightarrow Q$; $D \rightarrow P$
- (4) $A \rightarrow R$; $B \rightarrow S$; $C \rightarrow P$; $D \rightarrow Q$

Answer (1)

- Sol. The correct matches are
 - (A) Wurtz $\rightarrow R R$
 - (B) Fittig $\rightarrow \langle \bigcirc \rangle \langle \bigcirc \rangle$
 - (C) Wurtz fittig $\rightarrow \langle O \rangle R$
 - (D) Sandmeyer $\rightarrow \bigcirc$
- If volume of ideal gas is increased isothermally, then its internal energy
 - (1) Increased
 - (2) Remains constant
 - (3) Is decreased
 - (4) Can be increased or decreased

Answer (2)

- Sol. Internal energy of ideal gas depends only upon temperature.
- 13. Arrange the following ligands according to their increasing order of field strength

(1)
$$S^{2-} < CO < NH_3 < en < C_2O_4^{2-}$$

(2)
$$S^{2-} < NH_3 < en < CO < C_2O_4^{2-}$$

(3)
$$S^{2-} < C_2O_4^{2-} < NH_3 < en < CO$$

(4)
$$CO < en < NH_3 < C_2O_4^{2-} < S^{2-}$$

Sol. The correct order of field strength is

$$S^{2-} < C_2 O_4^{2-} < NH_3 < en < CO$$

14. Consider the following molecule

$$H_{B}$$
 $CO_{2}H_{A}$ H_{D}

Select the correct order of acidic strength

- (1) $H_A > H_D > H_B > H_C$ (2) $H_B > H_A > H_D > H_C$
- (3) $H_A > H_B > H_C > H_D$ (4) $H_C > H_B > H_D > H_A$

Answer (1)

Sol. The correct order of acidic strength is

 $H_A > H_D > H_B > H_C$

- 15. Which of the following compound is used as the antacid?
 - (1) Ranitidine
 - (2) Prontosil
 - (3) Norethindrone
 - (4) Codeine

Answer (1)

Sol. Ranitidine is used as the antacid.

- 16. The role of SiO₂ in Cu extraction is
 - (1) Converts FeO to FeSiO₃
 - (2) Converts CaO to CaSiO3
 - (3) Reduces Cu₂S to Cu
 - (4) None of these

Answer (1)

Sol. It converts FeO to FeSiO3

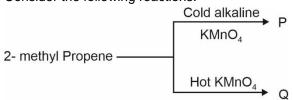
17. Assertion: Ketoses gives seliwanoff test.

Reason: Ketoses undergo β- elimination to form furfural.

- (1) Assertion and reason both are correct and reason is the correct explanation of assertion
- (2) Assertion and reason both are correct but reason is not the correct explanation of assertion.
- (3) Assertion is correct and reason is incorrect
- (4) Assertion is incorrect but reason is correct.

Answer (1)

- **Sol.** Assertion and reason both are correct and reason is the correct explanation of assertion.
- 18. Consider the following reactions:



The products P and Q respectively are?

Answer (2)

Sol.

(4) HCOOH and CH₃— C — CH₃

SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a **NUMERICAL VALUE.** For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 06.25, 07.00, -00.33, -00.30, 30.27, -27.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

21. For given cell, at T K

Pt
$$|H_2(g)|H^+||Fe^{3+}; Fe^{2+}|Pt$$

(1bar) (1M)

$$E_{cell} = .712 \text{ V}$$

$$E_{cell}^{\circ} = .770 \text{ V}$$

if
$$\frac{\left[Fe^{2+}\right]}{\left[Fe^{3+}\right]}$$
 is t $\left(\frac{2.303 \, RT}{F} = .058\right)$

then find
$$\left(\frac{t}{5}\right)$$

Answer (2)

Sol. .712 = .770 -
$$\frac{.058}{2} log \left[\frac{Fe^{2+}}{Fe^{3+}} \right]^2$$

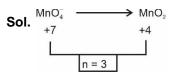
$$-.058 = -.058 log \frac{\left[Fe^{2+} \right]}{\left[Fe^{3+} \right]}$$

$$\frac{Fe^{2+}}{Fe^{3+}} = 10 = t$$

$$\frac{t}{5} = 2$$

22. How many moles of electrons are required to reduce 1 mole of permanganate ions into manganese dioxide

Answer (3)



3 mole of e- are required

 600 mL of 0.04 M HCl is mixed with 400 mL of 0.02 M H₂SO₄. Find out the pH of resulting solution (Nearest integer).

Answer (01.00)

Sol. m moles of H⁺ from HCl = 0.04×600

m moles of H⁺ from $H_2SO_4 = 0.02 \times 2 \times 400$

Total m moles of $H^+ = 24 + 16 = 40$

Final volume of solution = 1000 mL

$$[H^+] = \frac{40}{1000} = 0.04 \text{ M}$$

$$pH = - log 0.04 = 1.4$$

24. A solution of 2 g of a solute and 20 g water has boiling point 373.52 K. Then find the molar mass of solute in grams? [Given: K_b = 0.52 K kg/mole and solute is non-electrolyte].

Answer (100)

Sol. $\Delta T_b = K_b.m$

$$0.52 = 0.52 \times \frac{2/M}{02}$$

$$M = 100 g$$

25. When first order kinetic, rate constant is 2.011×10^{-3} sec⁻¹, the time taken in decomposition of substance from 7 g to 2 g will be. [Use log7 = 0.845 and log2 = 0.301]

Answer (623)

Sol. $A \rightarrow Products$

Initial moles of A = $\frac{7}{M}$ (M is molar mass of A)

Final moles of A =
$$\frac{2}{M}$$

Rate constant K = $2.011 \times 10^{-3} \text{ s}^{-1}$

$$t = \frac{2.303}{k} \log \frac{7}{2}$$
$$= \frac{2.303}{2.011 \times 10^{-3}} [0.845 - 0.301]$$

$$= 623 s$$

MATHEMATICS

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer:

- Coefficient of x^{301} in $(1 + x)^{500} + x(1 + x)^{499}$ + $x^2(1 + x)^{498}$ + + x^{500} is equal to
 - (1) $^{506}C_{306}$
 - $^{501}C_{300}$ (2)
 - (3) $^{501}C_{301}$
 - (4) $^{500}C_{300}$

Answer (3)

Sol. Coeff of
$$x^{301} = {}^{500}C_{301} + {}^{499}C_{300} + {}^{498}C_{299} + ... + {}^{199}C_{0}$$

$$= {}^{500}C_{199} + {}^{499}C_{199} + {}^{498}C_{199} + ... + {}^{199}C_{199}$$

$$= {}^{501}C_{200}$$

$$= {}^{501}C_{301}$$

- $\tan 15^{\circ} + \frac{1}{\tan 165^{\circ}} + \frac{1}{\tan 105^{\circ}} + \tan 195^{\circ} = 2a$ value of $\left(a + \frac{1}{a}\right)$ is
 - (1) $4-2\sqrt{3}$ (2) $\frac{-4}{\sqrt{3}}$

(3) 2

(4) $5 - \frac{3}{2}\sqrt{3}$

Answer (2)

Sol.
$$\tan 15^{\circ} + \cot 165^{\circ} + \cot 105^{\circ} + \tan 195^{\circ}$$

 $= \tan 15^{\circ} - \cot 15^{\circ} - \tan 15^{\circ} + \tan 15^{\circ}$
 $= \tan 15^{\circ} - \cot 15^{\circ}$
 $= -2\sqrt{3}$
 $\Rightarrow a = -\sqrt{3}$
 $a + \frac{1}{a} = -\sqrt{3} - \frac{1}{\sqrt{3}} = \frac{-4}{\sqrt{3}}$

If set $A = \{a, b, c\}$

 $R: A \rightarrow A$

 $R = \{(a,b), (b,c)\}$

How many elements should be added for making it symmetric and transitive.

(1) 2

(2) 3

(3) 4

(4) 7

Answer (4)

Sol. For symmetric

$$(a, b), (b, c) \in R$$

$$\Rightarrow$$
 (b, a), (c, b) $\in R$

For transitive.

$$(a, b), (b, c) \in R$$

$$\Rightarrow$$
 (a, c) $\in R$

Now,

$$(a, c) \in R$$

$$\Rightarrow$$
 $(c, a) \in R$ {For symmetric}

$$(a, b), (b, a) \in R$$

$$\Rightarrow$$
 (a, a) $\in R$

$$(b, c), (c, b) \in R$$

$$\Rightarrow$$
 $(b, b) \in R$

$$(c, b), (b, c) \in R$$

$$\Rightarrow$$
 $(c, c) \in R$

: elements to be added

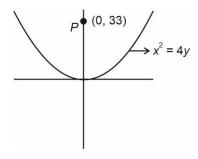
$$\{(b, a) (c, b) (b, b) (a, a) (a, c) (c, a) (c, c)\}$$

Total 7 elements

- Let P(h, k) be two points on $x^2 = 4y$ which is at 4. shortest distance from Q(0, 33) then difference of distances of P(h, k) from directrix of $y^2 = 4(x + y)$ is
 - (1) 2
 - (2) 4
 - (3) 6
 - (4) 8

Answer (2)

Sol. For normal through (0, 33)



Normal at point $(2t, \ell^2)$

$$x = -ty + 2at + at^3$$

$$0 = -t \cdot 33 + 2t + t^3$$

$$\Rightarrow$$
 $t = 0$ OR $\pm \sqrt{31}$

Points at which normal are drawn are

$$A(0, 0), B(2\sqrt{31}, 31), C(-2\sqrt{31}, 31)$$

Shortest distance

$$= PB = PC = \sqrt{124 + 4} = 8\sqrt{2}$$
 units

Given parabola $(y-2)^2 = 4(x+1)$

Directrix is x = -2, that is line L

$$B_L - C_L = |(-2 + 2\sqrt{31}) - (2 + 2\sqrt{31})|$$

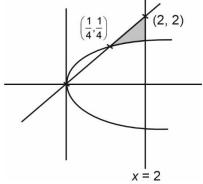
= 4

- Area bounded by larger part in I quadrant by $x = 4y^2$, x = 2 and y = x is A then 3A equals

 - (1) $6 + \frac{1}{32} 2\sqrt{2}$ (2) $2 + \frac{1}{96} \frac{2\sqrt{2}}{3}$
 - (3) $\frac{2\sqrt{2}}{3}$
- (4) 96

Answer (1)

Sol.



Shaded area is the required area

$$A = \int_{1/4}^{2} \left(x - \frac{\sqrt{x}}{2} \right) dx$$

$$=\frac{x^2}{2}-\frac{x^{3/2}}{3}\bigg|_{1/4}^2$$

$$= \left(2 - \frac{2\sqrt{2}}{3}\right) - \left(\frac{1}{32} - \frac{1}{24}\right)$$

$$=2+\frac{1}{96}-\frac{2\sqrt{2}}{3}$$

$$\Rightarrow$$
 3A = 6 + $\frac{1}{32}$ - 2 $\sqrt{2}$ sq. units.

- 6. A die with points (2, 1, 0, -1, -2, 3) is thrown 5 times. The probability that the product of outcomes on all throws is positive is
 - 521 2592
 - (2)

Answer (1)

Sol. Either all outcomes are positive or any two are negative.

The required probability = ${}^5C_5\left(\frac{1}{2}\right)^5 + {}^5C_2\left(\frac{1}{3}\right)^2\left(\frac{1}{2}\right)^3$

$$+{}^{5}C_{4}\left(\frac{1}{3}\right)^{4}\left(\frac{1}{2}\right)^{1} = \frac{5}{162} + \frac{1}{32} + \frac{5}{36} = \frac{521}{2592}$$

7. Let $S = \{1, 2, 3, 4, 5\}$

> if $f: S \to P(S)$, where P(S) is power set of S. Then number of one-one functions f can be made is

- $(1) (32)^5$

Answer (2)

Sol.
$$n(S) = 5$$

$$n(P(S)) = 2^5 = 32$$

$$\begin{array}{c}
S & P(S) \\
\begin{pmatrix}
x_1 \\
x_2 \\
x_3 \\
x_4 \\
x_5
\end{pmatrix} & \begin{pmatrix}
y_1 \\
y_2 \\
\vdots \\
\vdots \\
y_{32}
\end{pmatrix}$$

$$\therefore$$
 No. of one-one function= 32 x 31 x 30 x 24 x 28

$$=\frac{32!}{27!}$$

8. A line is cutting x axis and y axis at two points A and B, respectively, where OA = a, OB = b. A perpendicular is drawn from O (origin) to AB at an angle of $\frac{\pi}{6}$ from positive x-axis. If area of triangle

$$OAB = \frac{98\sqrt{3}}{3}$$
 sq. units, then $\sqrt{3} a + b$ is equal to

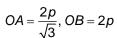
- (1) 28
- (2) 14
- (3) 12
- (4) 7

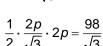
Answer (1)

Sol. Let the perpendicular distance of line from origin is

$$\Rightarrow$$
 Equation of $AB: \frac{x\sqrt{3}}{2} + \frac{y}{2} = p$

$$\Rightarrow \frac{x}{\frac{2p}{\sqrt{3}}} + \frac{y}{2p} = 1$$





$$\Rightarrow p = 7$$

$$OA = a = \frac{14}{\sqrt{3}}$$

$$OB = b = 14$$

$$\sqrt{3}a + b$$

$$\Rightarrow$$
 14 + 14 = 28

9. For solution of differential equation

$$\frac{dy}{dx} - \frac{3x^5 \tan^{-1}(x^3)}{(1+x^6)^{\frac{3}{2}}}y = -\frac{x^3 \tan^{-1} x^3}{\sqrt{1+x^6}}$$

given that y(0) = 0 then y(1) is

(1)
$$1-e^{\frac{\pi}{4\sqrt{2}}}$$

(2)
$$1 - e^{\left(\frac{1}{\sqrt{2}} - \frac{\pi}{4\sqrt{2}}\right)}$$

(3)
$$e^{\frac{1}{\sqrt{2}}} - e^{\frac{\pi}{4\sqrt{2}}}$$

(4)
$$e^{\frac{\pi}{4\sqrt{2}}}$$

Answer (2)

Sol. IF =
$$\int \frac{-3x^5 \tan^{-1}(x^3)}{(1+x^6)^{\frac{3}{2}}} dx$$

Let
$$\tan^{-1}(x^3) = t$$

$$\mathsf{IF} = e^{-\int t \sin t} = e^{\left(t \cos t - \sin t\right)}$$

Solution of Differential equation

$$y \cdot e^{(t\cos t - \sin t)} = \int e^{(t\cos t - \sin t)} (-t\sin t) dt$$

$$y \cdot e^{(t\cos t - \sin t)} = e^{(t\cos t - \sin t)} + c$$

$$t = 0 \rightarrow y = 0$$

$$\therefore c = -1$$

When
$$x = 1$$
, $t = \frac{\pi}{4}$

$$y \cdot e^{\left(\frac{\pi}{4\sqrt{2}} - \frac{1}{\sqrt{2}}\right)} = e^{\left(\frac{\pi}{4\sqrt{2}} - \frac{1}{\sqrt{2}}\right)} - 1$$

$$y = 1 - e^{\left(\frac{1}{\sqrt{2}} - \frac{\pi}{4\sqrt{2}}\right)}$$

10.
$$\frac{3(e-1)}{e} \int_{1}^{2} x^{2} e^{[x]+[x^{3}]} dx$$
 equals

- (1) = 9 = 1
- (2) $a^8 1$
- $(3) 0^8$
- (4) $e^9 e^9$

Answer (3)

Sol.
$$I = \int_{1}^{2} x^{2} e^{[x] + [x^{3}]} dx = e \int_{1}^{2} x^{2} \cdot e^{[x^{3}]} dx$$

Let
$$x^3 = t$$

$$I = e \int_{1}^{8} \frac{dt}{3} e^{[t]} = \frac{e}{3} (e + e^{2} + ... + e^{7})$$

$$=\frac{e^2}{3}\left(\frac{e^7-1}{e-1}\right)$$

So,
$$\frac{3(e-1)}{e} \cdot \frac{e^2}{3} \cdot \frac{e^7 - 1}{e-1} = e^8 - e$$

- 11. \hat{n} is a vector, $\vec{a} \neq 0$, $\vec{b} \neq 0$. If $\vec{n} \perp \vec{c}$, $\vec{a} = \alpha \vec{b} \hat{n}$ and $\vec{b} \cdot \vec{c} = 12$ then the value of $|\vec{c} \times (\vec{a} \times \vec{b})|$ equals (where \hat{n} represents unit vector in the direction of \vec{n})
 - (1) 144
 - (2) $\sqrt{12}$
 - (3) 12
 - (4) 24

Sol.
$$\vec{a} = \vec{\alpha} \vec{b} - \hat{n}$$

$$\Rightarrow \vec{a} \times \vec{b} = -\hat{n} \times \vec{b}$$

Now,

$$|\vec{c} \times (\vec{a} \times \vec{b})|$$

$$= |\vec{c} \times (-\hat{n} \times \vec{b})|$$

$$= |\hat{n}(12) - \vec{b}(0)|$$

= 12

SECTION - B

Numerical Value Type Questions: This section contains 10 questions. In Section B, attempt any five questions out of 10. The answer to each question is a **NUMERICAL VALUE.** For each question, enter the correct numerical value (in decimal notation, truncated/rounded-off to the second decimal place; e.g. 06.25, 07.00, -00.33, -00.30, 30.27, -27.30) using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

21.
$$\lim_{x\to 0} \frac{\int_{0}^{x} \frac{t^3}{1+t^6} dt}{x^4}$$
 equals

Answer (12)

Sol.
$$\lim_{x\to 0} \frac{48\int\limits_0^x \frac{t^3}{t^6+1} dt}{x^4}$$

As $\frac{0}{0}$ form, applying L' hospital rule we get

$$\lim_{x \to 0} 48 \frac{x^3}{\left(x^6 + 1\right) \cdot 4x^3} = 48 \cdot \frac{1}{4} = 12$$

22. If
$$a_n = \frac{-2}{4n^2 - 16n + 15}$$
 and $a_1 + a_2 + \dots + a_{25} = \frac{m}{n}$ where m and n are coprime, then the value of $m + n$ is

Answer (191)

Sol.
$$a_n = \frac{-2}{4n^2 - 16n + 15} = \frac{-2}{(2n - 3)(2n - 5)}$$

$$= \frac{1}{2n - 3} - \frac{1}{2n - 5}$$

$$a_1 + a_2 + \dots + a_{25} = \left(\frac{1}{-1} - \frac{1}{-3}\right) + \dots \left(\frac{1}{47} - \frac{1}{45}\right)$$

$$= \frac{1}{47} + \frac{1}{3} = \frac{50}{141}$$

$$m + n = 191$$

23. If
$$z = 1 + i$$
 and $z_1 = \frac{i + \overline{z}(1 - i)}{\overline{z}(1 - z)} = z_1$, then find the value of $\frac{12}{\pi} \arg(z_1)$.

Sol.
$$z_1 = \frac{i + \overline{z}(1-i)}{\overline{z}(1-z)} = \frac{i + (1-i)(1-i)}{(1-i)(-i)} = \frac{1}{1-i}$$

$$\arg z_1 = \arg\left(\frac{1}{1-i}\right) = -\arg(1-i) = \frac{\pi}{4}$$

$$\frac{12}{\pi}\arg(z_1) = \frac{12}{\pi} \times \frac{\pi}{4} = 3$$

24. Mean & Variance of 7 observations are 8 & 16 respectively, if number 14 is omitted then a & b are new mean & variance. The value of a + b is

Answer (19)

Sol. Let $x_1, \dots x_7$ are observation

New mean
$$=\frac{8 \times 7 - 14}{6} = 7$$

$$\sum_{i=1}^{n} x_i^2$$

$$\therefore \frac{i-1}{7} - 64 = 16 \Rightarrow \sum x_i^2 = 560$$

$$\sum x_{i(\text{new})}^2 = 560 - 14^2$$

$$\therefore b = \frac{364}{6} - 7^2 = \frac{70}{6} = \frac{35}{3}$$

$$\therefore a+b=7+\frac{35}{3}=\frac{56}{3}=18.67$$

Rounding off gives 19

25. If coefficient of x^{15} in expansion of $\left(ax^3 + \frac{1}{bx^{1/3}}\right)^{13}$ is equal to coefficient of x^{-15} in expansion of $\left(ax^{1/3} + \frac{1}{6x^3}\right)^{15}$ then |ab - 5| is equal to

Answer (04.00)

Sol.
$$a_n \left(ax^3 + \frac{1}{bx^{1/3}} \right)^{15} \Rightarrow T_{r+1} = {}^{15}C_r a^{15-r} \left(x^3 \right)^{15-r} b^{-r} x^{\frac{-r}{3}}$$

$$45 - 3r - \frac{r}{3} = 15 \Rightarrow \frac{10r}{3} = 30$$

$$\boxed{r = 9}$$

$$a_n \left(ax^{\frac{1}{3}} + \frac{1}{bx^3} \right)^{15} \Rightarrow T_{r+1} = {}^{15}C_r a^{15-r} x^{\frac{15-r}{3}} b^{-r} x^{-3r}$$

$$\frac{15-r}{3} - 3r = -15$$

$$15 - r - 9r = -45$$

$$\Rightarrow r = 6$$
So, ${}^{15}C_9 a^6 b^{-9} = {}^{15}C_6 a^9 b^{-6}$

$$\Rightarrow a^{-3} b^{-3} = 1$$

So,
$${}^{15}C_9a^6b^{-9} = {}^{15}C_6a^9b^{-6}$$

 $\Rightarrow a^{-3}b^{-3} = 1$
or $ab = 1$
 $|ab - 5| = 4$

Using 1, 2, 3, 5, 4-digit numbers are formed, where repetition is allowed. How many of them is divisible by 15?

Answer (21)

Sol. Units digit will be 5

For
$$(2, 2, 3) \Rightarrow \frac{3!}{2!} = 3$$

Linits digit will be 5

 $\frac{a}{a} + \frac{b}{c} + \frac{c}{5}$
 $\frac{5}{a+b+c} = (3\lambda + 1)$ type

For (a, b, c) possibilities are

For
$$(1, 1, 5) \Rightarrow \frac{3!}{2!} = 3$$

For
$$(1, 1, 2) \Rightarrow \frac{3!}{2!} = 3$$

For
$$(3, 3, 1) \Rightarrow \frac{3!}{2!} = 3$$

For
$$(5, 5, 3) \Rightarrow \frac{3!}{2!} = 3$$

For
$$(2, 3, 5) \Rightarrow 3! = 6$$

$$Total = 21$$

27. If $5f(x+y) = f(x) \cdot f(y)$ and f(3) = 320, then the value of f(1) is

Answer (20)

Sol.
$$5f(x+y) = f(x) \cdot f(y)$$
 ...(i) $f(3) = 320$

Put
$$x = 1$$
, $y = 2$ in (i)

$$5f(3) = f(1) \cdot f(2)$$

$$\Rightarrow f(1) \cdot f(2) = 5 \times 320 = 1600 \dots (ii)$$

Put
$$x = y = 1$$
 in (i)

$$5f(2) = (f(1))^2$$

$$\Rightarrow f(2) = \frac{(f(1))^2}{5} \qquad \dots (iii)$$

Using (iii) in (ii),

$$f(1) \cdot \frac{\left(f(1)\right)^2}{5} = 1600$$

$$(f(1))^3 = 8000$$

$$f(1) = 20$$

28. If for $\log_{\cos x}(\cot x) - 4\log_{(\sin x)}\cot x = 1$,

$$x = \sin^{-1}\left(\frac{\alpha + \sqrt{\beta}}{2}\right)$$
. Find $(\alpha + \beta)$, given $x \in \left(0, \frac{\pi}{2}\right)$

Answer (04.00)

Sol.
$$\log_{\cos x} \cot x - 4 \log_{\sin x} \cot x = 1$$

$$1 - \log_{\cos x} \sin x - 4(\log_{\sin x} \cos x - 1) = 1$$

Let
$$\log_{\cos x} \sin x = t$$

$$-t-4\left(\frac{1}{t}-1\right)=0$$

$$\Rightarrow t + \frac{4}{t} = 4$$

$$\Rightarrow t=2$$

$$\log_{\cos x} \sin x = 2$$

$$\Rightarrow \cos^2 x = \sin x$$

$$\Rightarrow 1 - \sin^2 x - \sin x = 0$$

$$\Rightarrow \sin^2 x + \sin x - 1 = 0$$

So,
$$\sin x = \frac{-1 \pm \sqrt{5}}{2}$$

$$\alpha = -1$$
, $\beta = 5$

$$\alpha + \beta = 4$$